Solve for convective film coefficient using: Principles of Heat Transfer by Kreith

Condition for nucleate boiling to occur

$$\Delta T_{n} := \left(\frac{8 \cdot \sigma \cdot h_{flux} \cdot T_{sat}}{\lambda \cdot \rho_{v} \cdot k_{liq}}\right)^{0.5} \qquad \Delta T_{n} = 1.672 \,\mathrm{K}$$

any differential film temp above this

Two Phase Convection-4.9mm Round Tube C3F8_5_12_08.xmcd

W.O. Miller iTi



${\bf h}_{\rm c}$ is the cooling from forced convection. The next step is to calculate the cooling from evaporation.

Solving for evaporation requires assuming a temperature difference between the tube wall and the saturation temperature of the fluid. After calculating this contribution we can solve for this temperature difference through the combination of both cooling mechanisms. If the end result agrees with the assumption, the process stops otherwise the process is repeated with a new assumed ΔT_{sat}

$$\operatorname{Re}_{TP_{i}} \coloneqq \frac{G1 \cdot (1 - x_{i}) \cdot D_{h}}{\mu_{liq}} \cdot (Ftt_{i})^{1.25} \cdot 10^{-4} \qquad \text{needed to calculate } S_{tt}$$

Two Phase Convection-4.9mm Round Tube C3F8_5_12_08.xmcd



For the assumed ΔT_{sat} (wall temp-fluid saturation temp) we calculate the value of ΔP_{sat} for C3F8. This value is the change in saturation pressure corresponding to the change in fluid temperature. At -25C the change in saturation pressure per degree C is 6800Pa, or for 5C ΔT_{sat} , the change is 34000Pa

 $\Delta T_{sat} \coloneqq 5.2 K$ temperature difference between wall and fluid, assumed to start the process.

$$\Delta p_{sat} \coloneqq \Delta T_{sat} \cdot 6800 \frac{Pa}{K} \qquad \Delta p_{sat} = 353.6 \cdot \text{mbar} \qquad \begin{array}{l} \text{pressure difference for temperature} \\ \text{difference} \end{array}$$

$$h_{b_0} \coloneqq .00122 \cdot \left(\frac{k_{liq}^{0.79} \cdot c_{liq}^{0.45} \cdot \rho_{liq}^{0.49}}{\sigma^{0.5} \cdot \mu_{liq}^{0.29} \cdot \lambda^{0.24} \cdot \rho_{v}^{0.24}} \right) \cdot \Delta T_{sat}^{0.24} \cdot \Delta p_{sat}^{0.75} \cdot \text{Stt}_{0} \qquad \begin{array}{c} \text{Chen} \end{array}$$

$$h_{b_0} = 1114.744 \cdot \frac{W}{m^2 \cdot K}$$
 $h_0 := h_{b_0} + h_{c_0}$ $h_0 = 1.736 \times 10^3 \cdot \frac{W}{m^2 \cdot K}$

$$\Delta t_{b_0} := \frac{h_{flux}}{h_0} \qquad \Delta t_{b_0} = 5.127 \, \text{K} \qquad \Delta T \text{ assumed was } 5.2 \text{C}$$

For the next quality 0.1

 $\Delta T_{\text{mat}} := 4.75 \text{K}$ temperature difference between wall and fluid, assumed to start the process.

$$\begin{split} & \Delta p_{sat} \coloneqq \Delta T_{sat} \cdot 6800 \frac{Pa}{K} \qquad \Delta p_{sat} = 323 \cdot mbar \qquad \begin{array}{l} & \text{pressure difference for temperature} \\ & h_{b_1} \coloneqq .00122 \cdot \left(\frac{k_{liq} \stackrel{0.79}{} \cdot c_{liq} \stackrel{0.45}{} \cdot \rho_{liq} \stackrel{0.49}{} \right) \\ & \sigma \stackrel{0.24}{} \cdot \rho_{v} \stackrel{0.24}{} \cdot \Delta p_{sat} \stackrel{0.75}{} \cdot Stt_1 \qquad \begin{array}{c} & \text{Chen} \\ & h_{b_1} = 1014.96 \cdot \frac{W}{m^2 \cdot K} \qquad & h_1 \coloneqq h_{b_1} + h_{c_1} \qquad & h_1 = 1.884 \times 10^3 \cdot \frac{W}{m^2 \cdot K} \\ & \Delta t_{b_1} \coloneqq \frac{h_{flux}}{h_1} \qquad & \Delta t_{b_1} = 4.725 \, K \qquad & \Delta T \text{ assumed was } 4.75 C \end{split}$$

Two Phase Convection-4.9mm Round Tube C3F8_5_12_08.xmcd For the next quality 0.3

<u>AT</u> = 3.75K temperature difference between wall and fluid, assumed to start the process.

$$\Delta p_{sat} \coloneqq \Delta T_{sat} \cdot 6800 \frac{Pa}{K} \qquad \Delta p_{sat} = 255 \cdot \text{mbar} \qquad \begin{array}{l} \text{pressure difference for temperature} \\ \text{difference} \end{array}$$

$$h_{b_2} \coloneqq .00122 \cdot \left(\frac{k_{liq} \stackrel{0.79}{-} \cdot c_{liq} \stackrel{0.45}{-} \cdot \rho_{liq} \stackrel{0.49}{-} \\ \sigma \stackrel{0.29}{-} \cdot \rho_{V} \stackrel{0.24}{-} \cdot \rho_{V} \stackrel{0.24}{-} \right) \cdot \Delta T_{sat} \stackrel{0.24}{-} \cdot \Delta p_{sat} \stackrel{0.75}{-} \cdot \text{Stt}_2 \qquad \begin{array}{c} \text{Chen} \end{array}$$

$$m_{b_2} = 792.307 \cdot \frac{10}{m^2 \cdot K}$$

 $m_2 = m_{b_2} + m_{c_2}$
 $m_2 = 2.307 \times 10^{-1} \cdot \frac{10}{m^2 \cdot K}$

$$\Delta t_{b_2} := \frac{n_{\text{flux}}}{h_2}$$
 $\Delta t_{b_2} = 3.761 \text{ K}$ ΔT assumed was 3.75C

For the next quality 0.5

<u>AT and state</u> = 3.25K temperature difference between wall and fluid, assumed to start the process.

$$\begin{split} & \bigtriangleup_{b_{3}} \coloneqq \Delta T_{sat} \cdot 6800 \frac{Pa}{K} \qquad \Delta p_{sat} = 221 \cdot mbar \qquad \begin{array}{l} & \text{pressure difference for temperature} \\ & h_{b_{3}} \coloneqq .00122 \cdot \left(\frac{k_{liq} \stackrel{0.79}{\longrightarrow} \cdot c_{liq} \stackrel{0.45}{\longrightarrow} \cdot \rho_{liq} \stackrel{0.49}{\longrightarrow} \right) \cdot \Delta T_{sat} \stackrel{0.24}{\longrightarrow} \cdot \Delta p_{sat} \stackrel{0.75}{\longrightarrow} \cdot Stt_{3} \qquad \begin{array}{l} & \text{Chen} \\ & h_{b_{3}} = 681.183 \cdot \frac{W}{m^{2} \cdot K} \qquad \qquad h_{3} \coloneqq h_{3} \Rightarrow h_{c_{3}} \qquad h_{3} = 2.739 \times 10^{3} \cdot \frac{W}{m^{2} \cdot K} \end{split}$$

 $\Delta t_{b_3} := \frac{h_{flux}}{h_3}$ $\Delta t_{b_3} = 3.25 \text{ K}$ ΔT assumed was 3.25C

For the next quality 0.6

AT set := 3.07K temperature difference between wall and fluid, assumed to start the process.

$$\Delta p_{sat} := \Delta T_{sat} \cdot 6800 \frac{Pa}{K} \qquad \Delta p_{sat} = 208.76 \cdot \text{mbar} \qquad \begin{array}{l} \text{pressure difference for temperature} \\ \text{difference} \end{array}$$

$$h_{b_4} := .00122 \cdot \left(\frac{k_{liq}}{\sigma^{0.5} \cdot \mu_{liq}} \cdot c_{liq}^{0.45} \cdot \rho_{liq}^{0.49}}{\sigma^{0.5} \cdot \mu_{liq}} \right) \cdot \Delta T_{sat}^{0.24} \cdot \Delta p_{sat}^{0.75} \cdot \text{Stt}_4$$

$$h_{b_4} := .00122 \cdot \left(\frac{k_{liq}}{\sigma^{0.5} \cdot \mu_{liq}} \cdot \lambda^{0.24} \cdot \rho_{V}^{0.24}}{\sigma^{0.5} \cdot \mu_{liq}} \right) \cdot \Delta T_{sat}^{0.24} \cdot \Delta p_{sat}^{0.75} \cdot \text{Stt}_4$$

$$h_{b_4} = 641.5 \cdot \frac{w}{m^2 \cdot K}$$

 $h_4 := h_{b_4} + h_{c_4}$
 $h_4 = 2.878 \times 10^3 \cdot \frac{w}{m^2 \cdot K}$

$$\Delta t_{b_4} := \frac{n_{\text{flux}}}{h_4}$$
 $\Delta t_{b_4} = 3.093 \text{ K}$ ΔT assumed was 3.07C

For the next quality 0.7

<u>AT set</u> := 2.97K temperature difference between wall and fluid, assumed to start the process.

$$\Delta p_{sat} \coloneqq \Delta T_{sat} \cdot 6800 \frac{Pa}{K} \qquad \Delta p_{sat} = 201.96 \cdot \text{mbar} \qquad \begin{array}{l} \text{pressure difference for temperature} \\ \text{difference} \end{array}$$

$$h_{b_{5}} \coloneqq .00122 \cdot \left(\frac{k_{liq} \stackrel{0.79}{-} \cdot c_{liq} \stackrel{0.45}{-} \cdot \rho_{liq} \stackrel{0.49}{-} \\ \sigma \stackrel{0.29}{-} \cdot \mu_{liq} \stackrel{0.29}{-} \cdot \lambda \stackrel{0.24}{-} \cdot \rho_{v} \stackrel{0.24}{-} \right) \cdot \Delta T_{sat} \stackrel{0.24}{-} \cdot \Delta p_{sat} \stackrel{0.75}{-} \cdot \text{Stt}_{5}$$

$$h_{b_5} = 619.155 \cdot \frac{W}{m^2 \cdot K}$$
 $h_5 := h_{b_5} + h_{c_5}$ $h_5 = 2.989 \times 10^3 \cdot \frac{W}{m^2 \cdot K}$

$$\Delta t_{b_5} := \frac{h_{flux}}{h_5}$$
 $\Delta t_{b_5} = 2.979 \,\mathrm{K}$ ΔT assumed was 2.97C

For the next quality 0.85

<u>AT</u> = 2.92K temperature difference between wall and fluid, assumed to start the process.

$$\Delta p_{sat} := \Delta T_{sat} \cdot 6800 \frac{Pa}{K} \qquad \Delta p_{sat} = 198.56 \cdot mbar \qquad \begin{array}{c} \text{pressure difference for temperature} \\ \text{difference} \end{array}$$

Two Phase Convection-4.9mm Round Tube C3F8_5_12_08.xmcd

W.O. Miller Convection Coefficient for Two Phase Flow with C3F8 at -25C LBNL-ATLAS iTi

$$\mathbf{h}_{b_{6}} \coloneqq .00122 \cdot \left(\frac{\mathbf{k}_{liq}^{0.79} \cdot \mathbf{c}_{liq}^{0.45} \cdot \rho_{liq}^{0.49}}{\sigma^{0.5} \cdot \mu_{liq}^{0.29} \cdot \lambda^{0.24} \cdot \rho_{v}^{0.24}} \right) \cdot \Delta \mathbf{T}_{sat}^{0.24} \cdot \Delta \mathbf{p}_{sat}^{0.75} \cdot \mathbf{Stt}_{6}$$

$$h_{b_6} = 608.024 \cdot \frac{W}{m^2 \cdot K}$$
 $h_6 := h_{b_6} + h_{c_6}$ $h_6 = 3.044 \times 10^3 \cdot \frac{W}{m^2 \cdot K}$

$$\Delta t_{b_6} := \frac{h_{\text{flux}}}{h_6} \qquad \Delta t_{b_6} = 2.925 \,\text{K}$$

 ΔT assumed was 2.92C